

**CAMBRIDGE INTERNATIONAL EXAMINATIONS**

Cambridge International General Certificate of Secondary Education

## **MARK SCHEME for the May/June 2015 series**

### **0654 CO-ORDINATED SCIENCES**

**0654/51**

Paper 5 (Practical), maximum raw mark 45

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Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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- 1 (a) outline concave on one side ;  
projections on the other side ; [2]

(b) (i)

test solution	observation
Benedict's solution	<b>green / yellow / orange / red ;</b>
biuret solution	<b>blue / no change / colour stays the same ;</b>
iodine solution	<b>brown / orange / no change / colour stays the same ;</b>

[3]

- (ii) reducing sugar / glucose ; [1]  
(**NOT** sugar. **DO NOT ALLOW** additional food groups)

- (c) (i) several small circles labelled 'stained' or 'coloured' or (c)(i) or red ;  
close to the outer ridged edge ; [2]



- (ii) xylem ;  
water transport (**ALLOW** water and any idea of movement, 'absorbs water' is not enough) [2]

- (d) star shape labelled 'transport tissue' or 'xylem' or ecf from (c)(ii) ;  
one structure in the middle ; [2]

- (e) different temperatures in separate experiments ;  
time for coloured water to appear at top of cut stem / set time and measure distance moved ;  
all other conditions / named condition kept constant ;  
(if one experiment proposed with gradual increase in temperature then can only score 2<sup>nd</sup> marking point) [3]

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- 2 (a) (i) white ppt./milky/cloudy white ; [1]
- (ii) blue/purple **AND** pH value in range 8–14 ; [1]
- (iii) calcium oxide / CaO ;  
(**ALLOW** quicklime / limewater / calcium hydroxide / Ca(OH)<sub>2</sub> ;  
**note**: accept answer if seen in (iv) [1]
- (iv) base / basic / alkali / alkaline ; [1]
- (b) (i) blue (and white) ppt. ; [1]
- (ii) blue (and white) ppt. ;  
(some) ppt dissolves soluble in excess (ammonia) ;  
to form darker blue (solution) ; [3]
- (iii) Cu<sup>2+</sup> / Cu(II) / copper (**not** Cu) ;  
copper oxide / CuO ; [2]  
**note**: both marks depend on 'blue' being reported in (b)(i) or (b)(ii)
- (c) (i) colourless ; [1]
- (ii) (faint) white ppt. / milky / cloudy white ;  
(ppt dissolves to form) colourless solution ; [2]  
(**DO NOT ALLOW** 'blue to colourless' for second marking point)
- (iii) chloride ; [1]
- (iv) Zn<sup>2+</sup> [1]

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- 3 (a)  $l_1$  recorded as whole number and clearly in mm ; [1]
- (b) (i)  $l_2$  recorded and  $e$  correctly calculated ; [1]  
**(DO NOT ALLOW** negative value of  $e$ )
- (ii) correct calculation of  $k$  ; [1]  
**(ALLOW** ecf from (b)(i), **ALLOW** 1 sig. fig. but then rounding must be correct)
- (c) all  $t$  values present and increasing ;  
**(ALLOW** 0:12 format)
- $T$  values correct minimum 2 sig. fig. ;  
 (if 0:12 format used for  $t$  then  $T = 12/20$  **NOT** 0.12/20)  
**(ALLOW** ecf from  $T$ )
- $T^2$  values correct **AND** to 2 sig. fig. ; [3]  
**(ALLOW** ecf from  $T$ )
- (d) (i) axes labelled with units ;  
 suitable choice of scales from (0,0) using at least half of each axis (m likely to be 0.1 per 2 cm) ;  
 at least three plots correct to  $\pm \frac{1}{2}$  small square ; ;  
 good best-fit straight line judgement ; [4]  
**(if non-linear** then do not award scale, plot or line marks **EXCEPT** if non-linear region is just between 0 and 0.2 kg, then do not award scale mark)
- (ii) indication on graph of how data obtained **AND**  $\geq$  half the line used ; [2]  
 correct calculation using data from graph ;
- (iii) correct calculation of  $k$  to 2/3 sig. fig. and correctly rounded ; [1]
- (e) **Method 1:** view at  $90^\circ$  to reading on scale / equivalent ;  
**Method 2:** Eye level with **top / bottom / middle / specified point** of oscillations / equivalent ; [2]